# Diphenyl Purine Derivatives as Peripherally Selective Cannabinoid Receptor 1 Antagonists 

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## (S) Supporting Information


#### Abstract

Cannabinoid receptor 1 (CB1) antagonists are potentially useful for the treatment of several diseases. However, clinical development of several CB1 antagonists was halted due to central nervous system (CNS)-related side effects including depression and suicidal ideation in some   users. Recently, studies have indicated that selective regulation of CB1 receptors in the periphery is a viable strategy for treating several important disorders. Past efforts to develop peripherally selective antagonists of CB1 have largely targeted rimonabant, an inverse agonist of CB1. Reported here are our efforts toward developing a peripherally selective CB1 antagonist based on the otenabant scaffold. Even though otenabant penetrates the CNS, it is unique among CB1 antagonists that have been clinically tested because it has properties that are normally associated with peripherally selective compounds. Our efforts have resulted in an orally absorbed compound that is a potent and selective CB1 antagonist with limited penetration into the CNS.


## INTRODUCTION

Cannabinoid receptors belong to the endocannabinoid (EC) system, which consists of receptors, transporters, endocannabinoids, and the enzymes involved in synthesis and degradation of endocannabinoids. To date, two receptors have been identified - CB1 and CB2. ${ }^{1}$ Both CB1 and CB2 are G protein-coupled receptors (GPCRs) primarily activating inhibitory $G$ proteins $\left(\mathrm{G}_{\mathrm{i} / \mathrm{o}}\right)$. ${ }^{2}$ Of the two, CB1 is prominently expressed in the central nervous system (CNS). ${ }^{1}$ However, it is also expressed peripherally in a number of peripheral tissues. The CB2 receptor is nominally expressed in the brain. However, it is highly expressed in cells of the immune system. ${ }^{3}$ The CB1 receptor is also a prominent target of drugs of abuse including ( - )- $\Delta^{9}$-tetrahydrocannabinol (THC), the main psychoactive constituent of marijuana. ${ }^{4}$

In recent years, CB1 antagonists have received attention in the treatment of disorders that have a central nervous system (CNS)-related craving component, including alcoholism. ${ }^{5}$ Further, CB1 is a validated drug target to treat obesity, metabolic syndrome, liver disease, diabetes, and dyslipidemias through both CNS (anorexic) and peripheral (metabolic) effects. ${ }^{6}$ Despite the therapeutic promise of CB1 antagonists, this drug class has been limited by adverse effects including anxiety and depression, and the first clinically approved CB1 antagonist (inverse agonist) for weight loss, rimonabant (1, SR141716A) (Figure 1), was withdrawn from Europe. ${ }^{7}$ Consequently, development or clinical trials of several CB1 antagonists, such as taranabant, otenabant (2), and ibipinabant, were halted due to regulatory concerns generated by $1 .{ }^{8}$

An alternative strategy to target this receptor is to develop antagonists that are peripherally restricted by virtue of not


1 (rimonabant)
(SR141716A)


2 (otenabant)
(CP-945,598)

Figure 1. Examples of CB1 antagonists.
being able to cross the blood-brain barrier (BBB) and thus avoiding CNS-mediated adverse effects noted with nontissue selective compounds. A similar strategy has been successful in the development of peripherally selective opioids. ${ }^{9}$ Several groups are currently pursuing this strategy. ${ }^{10}$ A few peripherally selective CB1 antagonists have been reported (Figure 2), and their further validation and characterization are underway. It should be noted almost all reported efforts (e.g., 3-7) at designing peripherally restricted antagonists of CB1 to date have focused on compounds that closely resemble 1.

Our group has been involved in the development of peripherally selective CB1 antagonists based on amide variations of compound 1 using two different approaches. First, charged compounds were produced based on the observation that such compounds cross the BBB only when

[^0]

4 (URB447)



Figure 2. CB1 antagonists that are reported to be selective for the periphery.
transported by a specific transporter. ${ }^{11}$ However, the compounds synthesized to date using this approach have low potency as reported in our previous publication. ${ }^{11 \mathrm{~b}}$ The second strategy involved generation of compounds with high topological polar surface areas (TPSAs). Past studies indicate that compounds with high TPSA, usually $>100$, have lower permeability into the CNS. ${ }^{12}$ This strategy was more successful and led to the identification of 7 that we have previously reported. ${ }^{13}$ While this compound was promising, having a brain to plasma ratio of less than $4 \%$ in a simple pharmacokinetic experiment, oral absorption of this compound was less than optimal. ${ }^{13}$

In this paper, we report our efforts toward the synthesis and characterization of peripherally selective CB1 antagonists based on 2. This compound is a highly selective CB1 receptor antagonist developed by Pfizer that was later abandoned due to CNS-related adverse effects during phase 3 clinical trials. This compound is unique among CB1 antagonists that have been clinically tested because it has properties that would be normally associated with a peripherally selective compound, including a TPSA of 102, three hydrogen bond donors, and formula weight $>500 .{ }^{14}$ However, 2 is CNS permeable. It was suggested that the intramolecular H -bonding observed in the X ray structure between the primary amide and the ethyl amine portion of the molecule effectively lowered the polarity of this compound, allowing for penetration into the CNS. ${ }^{14}$ Thus upon examination, this scaffold appeared to be more favorable for peripheralization as synthesis of compounds based on 2 but incapable of forming intramolecular H -bonding would be expected to be peripherally selective. This innovative approach to ligand design has now provided us with several promising compounds for further refinement.

## RESULTS

Approaches to Ligand Design and Testing. Synthesis, purification, and characterization of compounds are described in the Experimental Section. Antagonism of each compound toward CB1 was quantitatively tested using an in vitro calcium mobilization assay for CB1 as has been previously described, ${ }^{13,15}$ and this antagonism is expressed as an apparent dissociation affinity constant $\left(K_{\mathrm{e}}\right)$. Compounds that demonstrated $K_{e}<100 \mathrm{nM}$ were further characterized using competitive radioligand displacement of $\left[{ }^{3} \mathrm{H}\right]$ CP55940 at CB1 and CB2 overexpressing membrane preparations as described previously. ${ }^{13}$ Displacement data were quantified and expressed as an equilibrium dissociation constant $\left(K_{\mathrm{i}}\right)$ value at each receptor. Potent and selective compounds were progressed into an in vitro model of brain penetration in monolayers of MDCK-mdr1cells as described previously. ${ }^{11 \mathrm{~b}, 13}$ Certain compounds were also characterized for metabolic stability in human hepatic microsomal fractions (S9). Finally, in vivo testing was conducted in Sprague-Dawley (SD) rats to measure CNS permeability of these compounds.

Compound Development Strategy and Results. During our pursuit of peripherally selective CB1 antagonists, 8 (Figure 3) was synthesized and found to be highly potent,


9

Figure 3. Design of compound 9; a hybrid of compounds 2 and 8.
selective, and to have relatively low permeability in the MDCKmdr1 model. ${ }^{13}$ These results along with our interest in CB1 antagonists that possess a purine core, like 2 , moved us to make purine 9. Purine 9 is a potent $\left(K_{\mathrm{e}}=9 \mathrm{nM}\right)$ and selective (CB2/ CB1 selectivity ratio $\sim 3077$ ) hybrid of 2 and 8 with a TPSA of 90. Despite having a TPSA that is lower than that of 2,9 cannot form intramolecular hydrogen bonds, and therefore, was predicted to be more peripherally selective than 2 . This hypothesis was tested and confirmed (Table 1) using the MDCK-mdr1 permeability assay, ${ }^{13}$ where 9 was only $6 \%$

Table 1. Functional Assessment of Compound 9

| compd | TPSA | $\begin{gathered} \mathrm{K}_{\mathrm{e}} \\ \mathrm{CB} 1 \\ (\mathrm{nM}) \end{gathered}$ | $\begin{gathered} K_{\mathrm{i}} \mathrm{CB} 1 \\ (\mathrm{nM})\left[{ }^{3} \mathrm{H}\right] \\ \mathrm{CP} 55940 \end{gathered}$ | $\begin{gathered} K_{\mathrm{i}} \mathrm{CB} 2 \\ (\mathrm{nM})\left[{ }^{3} \mathrm{H}\right] \\ \mathrm{CP} 55940 \end{gathered}$ | CB2/ CB1 $K_{i}$ ratio | MDCKmdrl A to B (\%) ${ }^{a}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2 | 102 | 8.7 |  |  |  | 40 |
| 8 | 81 | 0.45 | 3.44 | 5504 | 1600 | 8 |
| 9 | 90 | 9 | 1.79 | 5507 | 3077 | 6 |

[^1]permeable (A-B), while 2 was $\sim 40 \%$ permeable (apical to basal, $A-B$ ) in the same assay. These positive results warranted further SAR exploration around 9 .

To further improve upon 9 , the amide was changed to more polar functionalities such as sulfonamide and urea (Figure 4,


Figure 4. Ligand design around compound 9, replacement of the amide functionality.

Table 2. Functional Assessment of Analogues of Compound 9

| compd | TPSA | $\begin{gathered} K_{\mathrm{e}} \\ \mathrm{CB1} \\ (\mathrm{nM}) \end{gathered}$ | $\begin{gathered} K_{\mathrm{i}} \mathrm{CB} 1 \\ (\mathrm{nM})\left[{ }^{3} \mathrm{H}\right] \\ \text { CP55940 } \end{gathered}$ | $\begin{gathered} K_{\mathrm{i}} \mathrm{CB} 2 \\ (\mathrm{nM})\left[{ }^{[3} \mathrm{H}\right] \\ \text { CP55940 } \end{gathered}$ | $\begin{aligned} & \text { CB2/ } \\ & \text { CB1 K } K_{i} \\ & \text { ratio } \end{aligned}$ | $\begin{aligned} & \text { MDCK- } \\ & \text { mdr1 A to } \\ & \text { B }(\%)^{a} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10 | 85 | 3.9 | 7.1 | 13947 | 1959 | <1 |
| 11 | 73 | 15.9 | 11 | 1657 | 151 | <1 |
| 12 | 101 | 2.9 | 6.2 | 948 | 153 | <1 |
| 13a | 88 | 19.2 | 25.5 | 20000 | 784 | <1 |
| 13b | 88 | 443 | 195 | 4645 | 24 |  |
| 13c | 88 | 46 | 337 | 8946 | 27 |  |
| 13d | 88 | 99 | 220 | 7779 | 35 |  |
| 13e | 88 | 62 | 269 | >20000 | >74 |  |
| 13f | 114 | 171 |  |  |  |  |
| 13g | 88 | 199 |  |  |  |  |

${ }^{a}$ Compound's permeability was measured from apical to basal (A to B) sides of the membrane.

Table 2). Carbamate $\mathbf{1 0}$ and amine $\mathbf{1 1}$ were first synthesized, the latter of which also serves as a common synthetic intermediate for other analogues. Interestingly, carbamate 10 was very potent ( $K_{\mathrm{e}}=3.9 \mathrm{nM}$ ), selective $(\mathrm{CB} 2 / \mathrm{CB} 1=1959)$, and poorly permeable in the MDCK-mdrl model (A to B $<1 \%$ ). Similarly, amine 11 was also very potent ( $K_{\mathrm{e}}=15.9 \mathrm{nM}$ ), selective $(C B 2: C B 1=151)$, and poorly permeable in the MDCK-mdr1 model (A to $\mathrm{B}<1 \%$ ). With amine 11 in hand, sulfonamide 12 and a series of ureas $13 \mathrm{a}-\mathrm{g}$ were synthesized. Sulfonamide 12 was especially interesting because it has high potency $\left(K_{\mathrm{e}}=2.9 \mathrm{nM}\right)$, selectivity $(\mathrm{CB} 2 / \mathrm{CB} 1=153)$, poor permeability in the MDCK-mdrl model (A to B is $<1 \%$ ), and a high TPSA (101), similar to 2 (TPSA = 102); it also lacks the
ability to hydrogen bond in an intramolecular fashion. Two members of structural series 13 were found to have interesting pharmacological profiles. Urea 13a was potent ( $K_{\mathrm{e}}=19.2 \mathrm{nM}$ ), selective $($ CB2/CB1 $=20000)$, and poorly permeable in the MDCK-mdrlassay ( A to B is $<1 \%$ ). Urea 13e was also found to be potent $\left(K_{\mathrm{e}}=62 \mathrm{nM}\right)$ and selective $(\mathrm{CB} 2 / \mathrm{CB} 1=74)$ as well.

The positive result observed with compounds $\mathbf{1 0 - 1 3 g}$ led to further exploration of the SAR around purines (Figure 5). The

14a, $\mathrm{R}=\mathrm{CO}_{2} t-\mathrm{Bu}$
15a, $\mathrm{R}=\mathrm{CO}_{2} t-\mathrm{Bu}$
14b, R=H
15b, R=H
14c, $\mathrm{R}=\mathrm{CONHEt}$
15c, R=CONHEt
14d, R=CONHi- Pr
15d, $\mathrm{R}=\mathrm{COMe}$
14e, $\mathrm{R}=\mathrm{CONH} n-\mathrm{Pr}$
14f, $\mathrm{R}=\mathrm{CONH} n-\mathrm{Bu}$
$14 \mathrm{~g}, \mathrm{R}=\mathrm{SO}_{2} \mathrm{Me}$


Figure 5. Exploration of different spacers.

4-amino-4-phenylpiperidine linker of compounds $\mathbf{1 0 - 1 3 g}$ was replaced with 4 - and 3 -aminopiperidine linkers to test the importance of the juxtaposition of the terminal polar groups of compounds $\mathbf{1 0} \mathbf{- 1 3 g}$. These changes also had the benefit of reducing the molecular weight of analogues that were synthesized by removing the phenyl group present in 1013g. Structural series $14,15,16$, and 17 were all made and evaluated pharmacologically. The data (Table 3) for these series show that the 4 -aminopiperidine linker of structural series 17 was superior to the linkers in structural series 14,15 , and 16 for all direct comparisons of the R group. However, in almost every case the 4 -amino-4-phenylpiperidine linker of 10$\mathbf{1 3 g}$ was more potent than its direct analogues in structural series 14-17.

However, several interesting compounds were identified in structural series 14-17. The tert-butoxycarbonyl (BOC) analogue 14a was potent ( $K_{\mathrm{e}}=92 \mathrm{nM}$ ) and selective (CB2/ CB1 = 568). In general, compounds from structural series 15 and 16 were weakly active against CB1 in the calcium mobilization assay and the compounds were not characterized further. Several analogues in structural series 17 were very interesting. The BOC analogue 17a was very potent ( $K_{\mathrm{e}}=11$ $\mathrm{nM})$ and selective $(\mathrm{CB} 2 / \mathrm{CB} 1=1142)$. The urea analogues 17c-f all showed good activity ( $K_{\mathrm{e}}$ ranging from 49 to 149 nM ). Selectivity (CB2/CB1) for ureas $\mathbf{1 7 c}-\mathrm{f}$ ranged from 5 to 175 fold. Amides and sulfonamides $\mathbf{1 7 g}-\mathrm{k}$ were generally active with 3 of the 5 analogues having $\mathrm{Ke}<10 \mathrm{nM}$. One of these

Table 3. Functional Assessment of Different Spacers and Functional Groups

| compd | TPSA | $\begin{gathered} \mathrm{K}_{\mathrm{e}} \mathrm{CB} 1 \\ (\mathrm{nM}) \end{gathered}$ | $\begin{aligned} & K_{i} \text { CB1 (nM) } \\ & {\left[{ }^{3} \mathrm{H}\right] \text { CP55940 }} \end{aligned}$ | $\begin{aligned} & K_{\mathrm{i}} \mathrm{CB} 2(\mathrm{nM}) \\ & {\left[{ }^{3} \mathrm{H}\right] \mathrm{CP} 55940} \end{aligned}$ | CB2/CB1 <br> $K_{\mathrm{i}}$ ratio |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 14a | 85 | 92 | 24 | 14000 | 568 |
| 14b | 68 | 5706 |  |  |  |
| 14c | 88 | 1148 |  |  |  |
| 14d | 88 | 495 |  |  |  |
| 14 e | 88 | 1964 |  |  |  |
| 14f | 88 | 71 | 3343 | >20000 | $>5$ |
| 14 g | 101 | 1645 |  |  |  |
| 15a | 85 | 326 |  |  |  |
| 15b | 68 | 5546 |  |  |  |
| 15c | 88 | 1500 |  |  |  |
| 15d | 76 | 920 |  |  |  |
| 15e | 101 | 346 |  |  |  |
| 16a | 85 | 3927 |  |  |  |
| 16b | 68 | 8406 |  |  |  |
| 16c | 88 | 1153 |  |  |  |
| 16d | 76 | 4449 |  |  |  |
| 16e | 101 | 6044 |  |  |  |
| 17a | 85 | 11 | 1.5 | 1713 | 1142 |
| 17c | 88 | 149 |  |  |  |
| 17d | 88 | 133 | 14 | 2453 | 175 |
| 17e | 88 | 132 | 28 | 153 | 5 |
| 17f | 88 | 49 | 25 | 1548 | 62 |
| 17 g | 76 | 631 |  |  |  |
| 17h | 101 | 71 | 29 | 1073 | 37 |
| 17i | 76 | 3.2 |  |  |  |
| 17j | 76 | 0.7 |  |  |  |
| 17k | 101 | 0.3 | 1.2 | 5865 | 4888 |

compounds, $\mathbf{1 7 k}$, was found to be highly selective $(\mathrm{CB} 2 / \mathrm{CB} 1=$ 4888).

In Vitro Metabolic Stability and In Vivo Evaluation of Brain Penetration. A small set of potent and selective compounds were progressed into in vitro models of metabolic stability (Table 4). Stability in human plasma and human

Table 4. Functional Assessment of Select Compounds in for in Vitro Metabolic Stability

| compd | TPSA | $\begin{gathered} K_{\mathrm{e}} \\ \mathrm{CB1} \\ (\mathrm{nM}) \end{gathered}$ | $\begin{gathered} \text { CB2/ } \\ \text { CB1 } \\ \text { CP55940 } \end{gathered}$ | $\begin{aligned} & \text { MDCK- } \\ & \text { mdr1 A to } \\ & \text { B (\%) } \end{aligned}$ | in vitro metabolic stability |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | $\begin{aligned} & \mathrm{S} 9 \% \\ & \text { remaining } \\ & 120 \text { min } \end{aligned}$ | plasma \% remaining 60 min |
| 9 | 90 | 0.45 | 3077 | 6 | >90 | $>90$ |
| 12 | 101 | 2.9 | 153 | <1 | >90 | >90 |
| 13a | 88 | 19.2 | 784 | <1 | 43 | >90 |

hepatic S9 fractions was measured to gauge potential metabolic liabilities of these compounds. All compounds were stable in plasma up to 1 h . The stabilities in S9 fractions for these compounds were more variable. All compounds displayed acceptable metabolic stabilities; however, 13a did show significant metabolism in S9 fraction and was deemed unsuitable for further development.

Several compounds showed excellent in vitro properties, including compounds $\mathbf{1 7 a}-\mathrm{k}$. Because of limited resources only sulfonamide 12 was advanced into an in vivo experiment to evaluate its brain penetration in SD rats upon oral dosing. Future studies are planned with additional compounds.

Sulfonamide 12 was dosed orally in SD rats at $10 \mathrm{mg} / \mathrm{kg}$, and both plasma and brain levels were analyzed out to 8 h post dose (Table 5). The compound showed slow absorption and

Table 5. Pharmacokinetic Analyses of 12

|  | dose <br> $\mathrm{mg} / \mathrm{kg}$ <br> oral | sacrifice <br> time $(\mathrm{min})$ | plasma conc. <br> $(\mathrm{ng} / \mathrm{mL})$ | brain conc. <br> $(\mathrm{ng} / \mathrm{mL})$ | brain/ <br> plasma <br> ratio |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1 2}$ | 10 | 30 | 752 | 38 | 0.05 |
|  |  | 60 | 767 | 58 | 0.08 |
|  |  | 120 | 1188 | 122 | 0.10 |
|  |  | 240 | 1653 | 184 | 0.11 |
|  |  | 480 | 914 | 89 | 0.10 |

did not reach a maximum concentration until 240 min post dose. Sulfonamide 12 also showed diminished brain penetration with the brain to plasma ratio ranging from 0.05 to 0.11 . It should be noted that unperfused brain tissues, which are expected to contain $2-4 \%$ blood, were examined. ${ }^{16}$ Thus, it is concluded that $\mathbf{1 2}$ has limited brain penetration. Furthermore, 12 demonstrated excellent oral absorption and favorable clearance upon oral dosing in rats. Thus, this compound may be suitable for further development.

Synthesis. Compound 9 (Scheme 1) was prepared from commercially available piperidine 19 and the readily available

## Scheme 1. Synthesis of Compound $9^{a}$


${ }^{a}$ Reagents and conditions: (a) 19, triethylamine, ethanol, $55{ }^{\circ} \mathrm{C}, 3$ days.
chloro-purine 18. ${ }^{14}$ Chloro-purine 18, piperidine 19, and triethylamine were heated in ethanol at $55{ }^{\circ} \mathrm{C}$ for 3 days yielding 9 in $63 \%$ yield. In a similar manner chloro-purine 18 was reacted with piperidine 20 and triethylamine in ethanol at $80^{\circ} \mathrm{C}$ for 16 h to yield 10 in $84 \%$ yield (Scheme 2). The BOC group of $\mathbf{1 0}$ was removed using $20 \%$ trifluoroacetic acid (TFA) in dichloromethane to yield the primary amine 11 in $99 \%$. Sulfonamide 12 was prepared by reacting 11, methanesulfonyl chloride, and triethylamine in tetrahydrofuran (THF) in $82 \%$ yield. The ureas $13 \mathrm{a}-\mathrm{g}$ were made by reacting amine 11, triethylamine, and the appropriate isocyanate in THF in yields ranging from 69 to $92 \%$.

Chloro-purine 18 was converted to 14a, 15a, and 16a in yields ranging from 86 to $95 \%$ by reaction with the appropriate amine and triethylamine in ethanol at $80^{\circ} \mathrm{C}$ for 16 h (Scheme 3 ). The BOC groups were removed from 14a, 15a, and 16a to form amines $\mathbf{1 4 b}, \mathbf{1 5 b}$, and $\mathbf{1 6 b}$ using $30 \%$ TFA in dichloromethane in yields ranging from 65 to $88 \%$. Amines $\mathbf{1 4 b}, \mathbf{1 5 b}$, and $\mathbf{1 6 b}$ were reacted with the appropriate isocyanate and triethylamine in THF to generate ureas $\mathbf{1 4 c}-\mathbf{f}, \mathbf{1 5 c}$, and $\mathbf{1 6 c}$ in yields ranging from 65 to $95 \%$. Amines $\mathbf{1 5 b}$ and 16b were reacted with acetic anhydride in pyridine to yield amides $\mathbf{1 5 d}$ and $\mathbf{1 6 d}$ in $87 \%$ and $81 \%$ respectively. Sulfonamides $\mathbf{1 4 g}$, $15 e$, and $16 e$ were made from $14 b, 15 b$, and $16 b$ by reaction

Scheme 2. Synthesis of Analogues of Compound $9^{a}$




12

${ }^{a}$ Reagents and conditions: (a) 20, triethylamine, ethanol, $80^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (b) $20 \%$ TFA in dichloromethane; (c) methanesulfonyl chloride, triethylamine, THF; (d) appropriate isocyanate, triethylamine, THF.
with methanesulfonyl chloride and triethylamine in THF in yields ranging from 51 to $91 \%$.

Chloro-purine 18 and piperidine 21 were heated in ethanol at $80^{\circ} \mathrm{C}$ for 16 h yielding 17 a in $99 \%$ yield (Scheme 4 ). The remaining chemistry, to generate $\mathbf{1 7 b} \mathbf{- k}$, was performed in a similar manner to reactions discussed above.

## DISCUSSION

CB1 receptor antagonism is a validated strategy to treat several important disorders including diabetes, obesity, metabolic syndromes, liver diseases, and dyslipidemias. However, nontissue selective antagonists of this receptor are not clinically useful due to adverse CNS effects. CB1 receptors are also expressed in peripheral organs and targeting of these peripheral receptors has produced promising data in experimental models of important diseases. Thus, several groups are trying to develop peripherally selective antagonists of CB1 for therapeutic use. While some progress has been made in this direction, currently disclosed CB1 antagonists that are peripherally selective need further improvement and validation prior to clinical use. Furthermore, most of these compounds are closely related to $\mathbf{1}$. In this report, we disclose our efforts to develop peripherally selective antagonists of CB1 based on 2, which we believe is a promising scaffold for optimization. This is because $\mathbf{2}$ has several chemical properties that would indicate poor brain penetration. However, intramolecular hydrogen bonding may effectively lower the polarity of 2 and this has been suggested by Griffith and co-workers. ${ }^{14}$ Thus, we hypothesized that optimized design and synthesis of compounds based on 2 that do not have the ability to form

Scheme 3. Synthesis of Compounds with Different Spacers and Functional Groups ${ }^{a}$

${ }^{a}$ Reagents and conditions: (a) triethylamine, ethanol, $80{ }^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (b) $30 \%$ TFA in dichloromethane; (c) appropriate isocyanate, triethylamine, THF, rt; (d) acetic anhydride, pyridine; (e) methanesulfonyl chloride, triethylamine, THF.
intramolecular hydrogen bonds could lead to CB1 antagonists with limited brain penetration.

Several potent and selective CB1 antagonists have been reported in this paper including nine compounds that have $K_{\mathrm{e}}<$ 100 nM and selectivity $>100$-fold versus CB2. Potency at CB1 has been obtained with a variety of functional groups on the terminus of the aminopiperidine in structures $10-17 \mathrm{k}$. Carbamates, amines, sulfonamides, ureas, and amides have all been well tolerated at this position. The 4 -amino- 4 -phenylpiperidine linker present in compounds $\mathbf{1 0 - 1 3 g}$ was the most potent linker noted in this study. 4-Aminopiperidine derivatives were more potent that 3 -aminopiperidine analogues. However, of the $1-3$ aminopiperidine compounds in structural series 15 and 16, the R enantiomers (15) were generally more potent at CB1 (four out of five examples) than the $S$ enantiomers (16) in head to head comparisons. Of the 4 -aminopiperidine linker compounds, lacking a phenyl group on the piperidine ring, in structural series 14 and 17 , the analogues where the piperidine nitrogen was bonded to the purine core (structural series 17)

Scheme 4. Synthesis of Compounds with Different Spacers and Functional Groups Continued ${ }^{a}$

${ }^{a}$ Reagents and conditions: (a) 21, ethanol, $80^{\circ} \mathrm{C}, 16 \mathrm{~h}$; (b) $30 \%$ TFA in dichloromethane; (c) appropriate isocyanate, triethylamine, THF; (d) appropriate anhydride, triethylamine, THF; (e) appropriate sulfonyl chloride, triethylamine, THF.
were favored over the isomers where the 4 -amino group was bonded to the core (structural series 14). Compounds with branched terminal R groups were found to be more selective for CB1 over CB2 than compounds with less sterically demanding R groups. This trend was especially prevalent in ureas $13 \mathrm{a}-\mathrm{d}$ and $17 \mathrm{c}-\mathrm{f}$. In both of these series, $13 \mathrm{a}-\mathrm{d}$ and $17 c-f$, the compounds with the most branched R group have the highest selectivity ratio for CB2/CB1. This selectivity is achieved not by gains in CB1 binding affinity, but by lowering the binding affinity for CB2. While 2 is an inverse agonist at CB1, it is currently not known if the compounds described in this publication act as inverse agonist or neutral antagonists at the present time. Those studies will be performed in the future.

The removal of the intramolecular hydrogen bond found in 2 allowed us to synthesize compounds with significantly lower permeability in the MDCK-mdrl assay. This can be demonstrated by comparing 2 and 12. While 2 and 12 have almost identical TPSAs ( 102 vs 101), $\mathbf{1 2}$ which lacks the ability to hydrogen bond in an intramolecular fashion has vastly reduced permeability in the MDCK-mdr1 assay. Thus, confirming Griffith and co-worker's hypothesis. ${ }^{14}$

Compounds 9, 12, and 13a were identified for further evaluation. All of these compounds showed reasonable stability to human plasma. Both 9 and $\mathbf{1 2}$ had good stability in human S9 fraction with over $90 \%$ of compound remaining after 120 min of incubation. The stability of 13a to S9 fraction was acceptable, but it did show significant metabolism in the assay. Compound $\mathbf{1 2}$ was advanced into in vivo models to assess its brain penetration. In a rat PK assay, $\mathbf{1 2}$ showed brain to plasma ratios ranging from 0.05 to 0.11 , indicating limited brain penetration. Generally, compounds that have $<10 \%$ brain penetration are considered peripherally selective. This is also a
significant reduction in brain penetration in comparison to 2 in SD rats, which has $\sim 33 \%$ brain exposure at $2-3 \mathrm{~h}$ postdose. ${ }^{17}$

In conclusion, through our studies, we have discovered novel analogues of 2 with vastly reduced brain penetration. Of the compounds reported, $\mathbf{1 2}$ had excellent potency, selectivity, and oral absorption in rats. This compound has low levels of brain penetration as well. This compound will be evaluated in proof of concept studies for diseases such as obesity, liver fibroses, and diabetes in the future. Furthermore, sulfonamide 12 will also serve as a starting point for refinement to further lower the brain penetration of future compounds, if necessary. However, any such endeavor must be dictated by the proposed use, and an acceptable balance between therapeutic benefits and undesirable effects must be achieved. For example, antiobesity drugs should have a very favorable profile as far as adverse effects are concerned. By contrast, in other diseases such as liver disease or diabetes, a more permissive profile of adverse effects could be tolerated.

## EXPERIMENTAL SECTION

Compound Synthesis and Characterization. Chemistry General. Purity and characterization of compounds were established by a combination of HPLC, TLC, and NMR analytical techniques described below. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{CNMR}$ spectra were recorded on a Bruker Avance DPX-300 ( 300 MHz ) spectrometer and were determined in $\mathrm{CHCl}_{3}-d$ or $\mathrm{MeOH}-d_{4}$ with tetramethylsilane (TMS) ( 0.00 ppm ) or solvent peaks as the internal reference unless otherwise noted. Chemical shifts are reported in ppm relative to the solvent signal, and coupling constant $(J)$ values are reported in hertz $(\mathrm{Hz})$. Thin-layer chromatography (TLC) was performed on EMD precoated silica gel 60 F254 plates, and spots were visualized with UV light or $\mathrm{I}_{2}$ detection. Low-resolution mass spectra were obtained using a Waters Alliance HT/Micromass ZQ system (ESI). All test compounds were greater than $95 \%$ pure as determined by HPLC on an Agilent 1100 system using an Agilent Zorbax SB-Phenyl, $2.1 \times 150 \mathrm{~mm}, 5 \mu \mathrm{~m}$ column with gradient elution using the mobile phases (A) $\mathrm{H}_{2} \mathrm{O}$ containing $0.05 \%$ $\mathrm{CF}_{3} \mathrm{COOH}$ and (B) methanol. A flow rate of $1.0 \mathrm{~mL} / \mathrm{min}$ was used. All compounds were isolated as white or off-white solids.

1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phe-nylpiperidine-4-carboxamide (9). To a solution of 6-chloro-8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purine (9) ( $23.1 \mathrm{mg}, 0.062$ mmol, 1 equiv) in 2 mL of ethanol was added 4-carbamoyl-4-phenylpiperidin-1-ium trifluoroacetate $(19)(25 \mathrm{mg}, 0.123 \mathrm{mmol}, 2$ equiv) and triethylamine ( $0.03 \mathrm{~mL}, 0.19 \mathrm{mmol}, 3$ equiv). The reaction was heated to $55^{\circ} \mathrm{C}$ for 3 days. The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 21 mg $(63 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}\right.$, methanol- $\left.d_{4}\right) \delta \mathrm{ppm}$ 2.02-2.21 (m, 2 H) 2.62 (d, $J=13.37 \mathrm{~Hz}, 2 \mathrm{H}) 3.82$ (br. s., 2 H$) 4.98$ $(\mathrm{d}, J=15.54 \mathrm{~Hz}, 2 \mathrm{H}) 7.07-7.67(\mathrm{~m}, 13 \mathrm{H}) 8.02-8.29(\mathrm{~m}, 1 \mathrm{H}),[\mathrm{M}$ $+\mathrm{H}]^{+} 543.6$.
tert-Butyl N-\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}carbamate (10). To a solution of 18 (50 $\mathrm{mg}, 0.133 \mathrm{mmol}, 1$ equiv) in 2 mL of ethanol was added tert-butyl N -(4-phenylpiperidin-4-yl)carbamate (20) (44 mg, $0.16 \mathrm{mmol}, 1.2$ equiv) and triethylamine ( $0.03 \mathrm{~mL}, 0.20 \mathrm{mmol}, 1.5$ equiv). The reaction was heated to $80^{\circ} \mathrm{C}$ for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 69 mg ( $84 \%$ ) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm}$ 1.31-1.49 (m, 9 H) 2.09-2.28 (m, 2 H) 2.43 (br. s., 2 H) 3.62 (br. s., 2 H) 4.96 (s, 1 H) 5.30 (br. s., 2 H) $7.03-7.61$ (m, 13 H$) 8.40(\mathrm{~s}, 1$ H), $[\mathrm{M}+\mathrm{H}]^{+} 615.4$.

1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phe-nylpiperidin-4-amine (11). A solution of 10 ( $2.65 \mathrm{~g}, 4.3 \mathrm{mmol}$ ) was stirred in dichloromethane $(32 \mathrm{~mL})$ and trifluoroacetic acid $(8 \mathrm{~mL})$ for 1.5 h . The reaction was concentrated in vacuo. The crude product was dissolved in ethyl acetate and washed with 3.8 M NaOH . The aqueous
layer was extracted twice with ethyl acetate. The combined organic layer was washed with brine and dried with $\mathrm{MgSO}_{4}$ to yield 2.21 g ( $99 \%$ ) of pure desired product. 1H NMR ( 300 MHz , chloroform-d) $\delta$ ppm $1.87(\mathrm{~d}, J=13.56 \mathrm{~Hz}, 2 \mathrm{H}) 2.15-2.39(\mathrm{~m}, 2 \mathrm{H}) 3.82-4.11(\mathrm{~m}, 2$ H) 5.10 (br. s., 2 H$) 7.11-7.62(\mathrm{~m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$ 515.8.

N-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}methanesulfonamide (12). To a solution of 11 ( $8.4 \mathrm{mg}, 0.016 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added methanesulfonyl chloride $(0.01 \mathrm{~mL})$ and triethylamine $(0.02 \mathrm{~mL})$. The mixture was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield $8 \mathrm{mg}(82 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta$ ppm 2.22 (s, 3 H ) 2.30-2.44 (m, 2 H) 2.47-2.66 (m, 2 H ) 4.20 (br. s., 2 H ) 4.75 ( s, 2 H) $7.08-7.64(\mathrm{~m}, 13 \mathrm{H}) 8.38(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 593.3$.

General Procedure for Making Ureas from 1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-amine (11). To a solution of 11 ( $19 \mathrm{mg}, 0.036 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.015 \mathrm{~mL}, 0.108 \mathrm{mmol}, 3$ equiv) and the appropriate isocyanate ( 1.5 equiv). The reaction was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield pure compound.

3-tert-Butyl-1-\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}urea (13a). Reaction proceeded in $81 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.12$ (s, 9 H ) 2.13-2.27 (m, 2 H ) 2.27-2.43 (m, 2 H ) 3.67 (br. s., 2 H ) 3.96 (br. s., $1 \mathrm{H}) 4.78$ (br. s., 1 H$) 5.10-5.68(\mathrm{~m}, 2 \mathrm{H}) 7.07-7.63(\mathrm{~m}, 13 \mathrm{H}) 8.40$ (s, 1 H ), $[\mathrm{M}+\mathrm{H}]^{+} 614.7$.

1-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}-3-ethylurea (13b). Reaction proceeded in $81 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform-d) $\delta \mathrm{ppm} 0.85-1.01(\mathrm{~m}, 3 \mathrm{H})$ 2.11-2.29 (m, 2 H) 2.30-2.47 (m, 2 H) 2.96-3.20 (m, 2 H) 3.64 (br. s., 2 H$) 4.10-4.24(\mathrm{~m}, 1 \mathrm{H}) 5.00(\mathrm{~s}, 1 \mathrm{H}) 5.32(\mathrm{~d}, J=14.41 \mathrm{~Hz}, 2 \mathrm{H})$ $7.10-7.60(\mathrm{~m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 586.8$.

1-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}-3-(propan-2-yl)urea (13c). Reaction proceeded in $79 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.94$ (d, $J=$ $6.50 \mathrm{~Hz}, 6 \mathrm{H}) 2.13-2.29(\mathrm{~m}, 2 \mathrm{H}) 2.29-2.42(\mathrm{~m}, 2 \mathrm{H}) 3.51-3.76(\mathrm{~m}$, $2 \mathrm{H}) 3.95(\mathrm{~d}, J=7.82 \mathrm{~Hz}, 1 \mathrm{H}) 4.93(\mathrm{~s}, 1 \mathrm{H}) 5.32(\mathrm{~d}, J=13.56 \mathrm{~Hz}, 2$ H) $7.07-7.63(\mathrm{~m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 600.7$.

1-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}-3-propylurea (13d). Reaction proceeded in $88 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.64-0.77$ $(\mathrm{m}, 3 \mathrm{H}) 1.19-1.37(\mathrm{~m}, 2 \mathrm{H}) 2.10-2.27(\mathrm{~m}, 2 \mathrm{H}) 2.29-2.45(\mathrm{~m}, 2$ H) 2.87-3.08 (m, 2 H) 3.63 (br. s., 2 H) 4.28 (br. s., 1 H) 5.11 (s, 1 H) 5.36 (br. s., 2 H$) 7.10-7.60(\mathrm{~m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$ 600.5.

3-Butyl-1-\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}urea (13e). Reaction proceeded in $82 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform- $d$ ) $\delta \mathrm{ppm} 0.74-0.86(\mathrm{~m}, 3 \mathrm{H})$ $1.11(\mathrm{dq}, J=14.79,7.22 \mathrm{~Hz}, 2 \mathrm{H}) 1.20-1.31(\mathrm{~m}, 2 \mathrm{H}) 2.11-2.27(\mathrm{~m}$, $2 \mathrm{H}) 2.29-2.41(\mathrm{~m}, 2 \mathrm{H}) 3.05(\mathrm{q}, J=6.59 \mathrm{~Hz}, 2 \mathrm{H}) 3.63$ (br. s., 2 H ) $4.22(\mathrm{t}, J=5.13 \mathrm{~Hz}, 1 \mathrm{H}) 5.07(\mathrm{~s}, 1 \mathrm{H}) 5.34$ (br. s., 2 H$) 7.12-7.56$ $(\mathrm{m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 614.7$.

Ethyl 2-[(\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}carbamoyl)amino]acetate (13f). Reaction proceeded in $92 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm}$ $1.18-1.28(\mathrm{~m}, 3 \mathrm{H}) 2.11-2.29(\mathrm{~m}, 2 \mathrm{H}) 2.33-2.52(\mathrm{~m}, 2 \mathrm{H}) 3.64$ (br. s., 2 H$) 3.82-3.92(\mathrm{~m}, 2 \mathrm{H}) 4.09-4.24(\mathrm{~m}, 2 \mathrm{H}) 5.13$ (br. s., 1 H$)$ $5.23-5.52(\mathrm{~m}, 2 \mathrm{H}) 5.60(\mathrm{~s}, 1 \mathrm{H}) 7.05-7.59(\mathrm{~m}, 13 \mathrm{H}) 8.39(\mathrm{~s}, 1 \mathrm{H})$, $[\mathrm{M}+\mathrm{H}]^{+} 644.4$.

3-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-4-phenylpiperidin-4-yl\}-1-cyclohexylurea (13g). Reaction proceeded in $69 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.77-1.13$ (m, $3 \mathrm{H}) 1.15-1.36(\mathrm{~m}, 3 \mathrm{H}) 1.36-1.54(\mathrm{~m}, 3 \mathrm{H}) 1.56-1.76(\mathrm{~m}, 1 \mathrm{H})$ $2.12-2.28(\mathrm{~m}, 2 \mathrm{H}) 2.29-2.42(\mathrm{~m}, 2 \mathrm{H}) 3.34-3.55(\mathrm{~m}, 1 \mathrm{H}) 3.65$ (br. s., 2 H) $3.96-4.20(\mathrm{~m}, 1 \mathrm{H}) 4.96$ (s, 1 H$) 5.34$ (br. s., 2 H ) $7.06-7.61$ $(\mathrm{m}, 13 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 640.5$.
tert-Butyl 4-\{[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yllamino\}piperidine-1-carboxylate (14a). To a solution of 18 (151 $\mathrm{mg}, 0.403 \mathrm{mmol}, 1$ equiv) in 4 mL of ethanol was added tert-butyl 4 -aminopiperidine-1-carboxylate ( $96 \mathrm{mg}, 0.483 \mathrm{mmol}, 1.2$ equiv) and triethylamine ( $0.08 \mathrm{~mL}, 0.6 \mathrm{mmol}, 1.5$ equiv). The reaction was heated to $80^{\circ} \mathrm{C}$ for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-$ $100 \%$ ethyl acetate/hexanes to yield 203 mg (94\%) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm} 1.38-1.55(\mathrm{~m}, 11 \mathrm{H}) 2.14$ (d, $J=12.34 \mathrm{~Hz}, 2 \mathrm{H}) 2.98(\mathrm{t}, J=12.10 \mathrm{~Hz}, 2 \mathrm{H}) 4.00-4.21(\mathrm{~m}, 2 \mathrm{H})$ $4.28-4.52(\mathrm{~m}, 1 \mathrm{H}) 5.83(\mathrm{~d}, J=7.63 \mathrm{~Hz}, 1 \mathrm{H}) 7.17-7.27(\mathrm{~m}, 2 \mathrm{H})$ $7.31-7.44(\mathrm{~m}, 5 \mathrm{H}) 7.50(\mathrm{~d}, J=6.88 \mathrm{~Hz}, 1 \mathrm{H}) 8.44(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+$ $\mathrm{H}]^{+} 539.4$.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)-N-(piperidin-4-yl)-9H-purin-6-amine (14b). A solution of 14a ( $180 \mathrm{mg}, 0.33 \mathrm{mmol}$ ) was stirred in dichloromethane $(9 \mathrm{~mL})$ and trifluoroacetic acid $(4 \mathrm{~mL})$ for 16 h . The reaction was concentrated in vacuo. The crude product was dissolved in ethyl acetate and washed with saturated $\mathrm{NaHCO}_{3}$. The aqueous layer was extracted twice with ethyl acetate. The combined organic layer was washed with brine and dried with $\mathrm{MgSO}_{4}$. The crude material was purified by silica gel column chromatography using $0-$ 100\% CMA 80 ( $80 \%$ chloroform, $18 \%$ methanol, and $2 \%$ ammonium hydroxide)/ethyl acetate to yield 129 mg ( $88 \%$ ) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm} 1.45$ ( $\mathrm{qd}, J=11.66,3.81$ $\mathrm{Hz}, 2 \mathrm{H}) 2.10(\mathrm{~d}, J=11.87 \mathrm{~Hz}, 2 \mathrm{H}) 2.64-2.88(\mathrm{~m}, 2 \mathrm{H}) 3.10(\mathrm{~d}, J=$ $12.62 \mathrm{~Hz}, 2 \mathrm{H}) 4.27$ (br. s., 1 H$) 5.80(\mathrm{~d}, J=6.31 \mathrm{~Hz}, 1 \mathrm{H}) 7.01-7.55$ $(\mathrm{m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 439.6$.

General Procedure for Making Ureas from 8-(2-chlorophenyl)-9-(4-chlorophenyl)- N -(piperidin-4-yl)-9H-purin-6-amine 14b. To a solution of $\mathbf{1 4 b}(18.4 \mathrm{mg}, 0.042 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.02 \mathrm{~mL}, 0.126 \mathrm{mmol}, 3$ equiv) and the appropriate isocyanate ( 1.5 equiv). The reaction is stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using 0-100\% CMA 80/ ethyl acetate to yield pure compound.

4-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}-$N$-ethylpiperidine-1-carboxamide (14c). Reaction proceeded in 86\% yield. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform- $d) \delta \mathrm{ppm} 1.09(\mathrm{t}, J=7.21 \mathrm{~Hz}$, $3 \mathrm{H}) 1.34-1.60(\mathrm{~m}, 2 \mathrm{H}) 2.11(\mathrm{~d}, J=12.34 \mathrm{~Hz}, 2 \mathrm{H}) 2.83-3.06(\mathrm{~m}, 2$ H) $3.13-3.32(\mathrm{~m}, 2 \mathrm{H}) 3.92(\mathrm{~d}, J=13.37 \mathrm{~Hz}, 2 \mathrm{H}) 4.40(\mathrm{~d}, J=4.90$ $\mathrm{Hz}, 2 \mathrm{H}) 5.80(\mathrm{~d}, J=7.82 \mathrm{~Hz}, 1 \mathrm{H}) 7.04-7.51(\mathrm{~m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H})$, $[\mathrm{M}+\mathrm{H}]^{+} 510.4$.

4-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}-$N$-(propan-2-yl)piperidine-1-carboxamide (14d). Reaction proceeded in $91 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm}$ $1.18(\mathrm{~d}, J=6.50 \mathrm{~Hz}, 6 \mathrm{H}) 1.46-1.68(\mathrm{~m}, 2 \mathrm{H}) 2.19(\mathrm{~d}, J=10.55 \mathrm{~Hz}, 2$ H) $3.03(\mathrm{t}, J=11.59 \mathrm{~Hz}, 2 \mathrm{H}) 3.89-4.10(\mathrm{~m}, 3 \mathrm{H}) 4.29(\mathrm{~d}, J=7.16$ Hz, 1 H) 4.42 (br. s., 1 H$) 5.88(\mathrm{~d}, J=7.82 \mathrm{~Hz}, 1 \mathrm{H}) 7.13-7.64(\mathrm{~m}, 8$ H) $8.45(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 524.7$.

4-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}-$N$-propylpiperidine-1-carboxamide (14e). Reaction proceeded in $95 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.86-1.01$ $(\mathrm{m}, 3 \mathrm{H}) 1.44-1.67(\mathrm{~m}, 4 \mathrm{H}) 2.19(\mathrm{~d}, J=10.64 \mathrm{~Hz}, 2 \mathrm{H}) 3.05(\mathrm{t}, J=$ $11.73 \mathrm{~Hz}, 2 \mathrm{H}) 3.22(\mathrm{q}, J=6.59 \mathrm{~Hz}, 2 \mathrm{H}) 3.92-4.08(\mathrm{~m}, 2 \mathrm{H}) 4.43$ (br. s., 1 H$) 4.54(\mathrm{t}, J=5.04 \mathrm{~Hz}, 1 \mathrm{H}) 5.88(\mathrm{~d}, J=7.72 \mathrm{~Hz}, 1 \mathrm{H}) 7.09-$ $7.58(\mathrm{~m}, 8 \mathrm{H}) 8.45(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 524.8$.

N-Butyl-4-\{[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}piperidine-1-carboxamide (14f). Reaction proceeded in $91 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm} 0.79-0.92$ (m, $3 \mathrm{H}) 1.29(\mathrm{dq}, J=14.75,7.20 \mathrm{~Hz}, 2 \mathrm{H}) 1.36-1.58(\mathrm{~m}, 4 \mathrm{H}) 2.11(\mathrm{~d}, J$ $=10.64 \mathrm{~Hz}, 2 \mathrm{H}) 2.96(\mathrm{t}, J=11.73 \mathrm{~Hz}, 2 \mathrm{H}) 3.17(\mathrm{q}, J=6.75 \mathrm{~Hz}, 2 \mathrm{H})$ $3.81-4.01(\mathrm{~m}, 2 \mathrm{H}) 4.22-4.51(\mathrm{~m}, 2 \mathrm{H}) 5.79(\mathrm{~d}, J=7.54 \mathrm{~Hz}, 1 \mathrm{H})$ $7.01-7.53(\mathrm{~m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 538.4$.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)- N -(1-methanesulfonylpi-peridin-4-yl)-9H-purin-6-amine (14g). To a solution of $\mathbf{1 4 b}(18.4 \mathrm{mg}$, $0.042 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added methanesulfonyl chloride ( $0.005 \mathrm{~mL}, 0.063 \mathrm{mmol}, 1.5$ equiv) and triethylamine ( 0.02 $\mathrm{mL}, 0.126 \mathrm{mmol}, 3$ equiv). The mixture was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/
hexanes to yield 20 mg (91\%) of desired product. ${ }^{1} \mathrm{H}$ NMR (300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.58-1.76(\mathrm{~m}, 2 \mathrm{H}) 2.21(\mathrm{~d}, J=10.64 \mathrm{~Hz}$, $2 \mathrm{H}) 2.76(\mathrm{~s}, 3 \mathrm{H}) 2.90(\mathrm{t}, J=10.93 \mathrm{~Hz}, 2 \mathrm{H}) 3.77(\mathrm{~d}, J=12.15 \mathrm{~Hz}, 2$ H) 4.33 (br. s., 1 H) 5.84 (br. s., 1 H) $7.04-7.51$ (m, 8 H$) 8.36$ (s, 1 H), $[\mathrm{M}-\mathrm{H}]^{-}$515.7.
tert-Butyl (3R)-3-\{[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}piperidine-1-carboxylate (15a). To a solution of 18 ( $100 \mathrm{mg}, 0.267 \mathrm{mmol}, 1$ equiv) in 3 mL of ethanol was added tertbutyl (3R)-3-aminopiperidine-1-carboxylate ( $64 \mathrm{mg}, 0.32 \mathrm{mmol}, 1.2$ equiv) and triethylamine ( $0.06 \mathrm{~mL}, 0.4 \mathrm{mmol}, 1.5$ equiv). The reaction was heated to $80^{\circ} \mathrm{C}$ for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 124 mg ( $86 \%$ ) of desired product. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform- $d$ ) $\delta$ ppm 1.33 (br. s., 9 H$)$ $1.49-1.67(\mathrm{~m}, 2 \mathrm{H}) 1.74(\mathrm{~d}, J=6.97 \mathrm{~Hz}, 1 \mathrm{H}) 2.00(\mathrm{~d}, J=2.83 \mathrm{~Hz}, 1$ H) 3.20 (br. s., 2 H) 3.50 (br. s., 1 H) 3.88 (br. s., 1 H) 4.32 (br. s., 1 H) $5.87(\mathrm{~d}, J=7.44 \mathrm{~Hz}, 1 \mathrm{H}) 7.02-7.54(\mathrm{~m}, 8 \mathrm{H}) 8.38(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+$ $\mathrm{H}]^{+}$539.3.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)-N-[(3R)-piperidin-3-yl]-9H-purin-6-amine (15b). A solution of tert-butyl 15a (109 mg, 0.202 mmol ) was stirred in dichloromethane $(7 \mathrm{~mL})$ and trifluoroacetic acid $(3 \mathrm{~mL})$ for 16 h . The reaction was concentrated in vacuo. The crude product was dissolved in ethyl acetate and washed with saturated $\mathrm{NaHCO}_{3}$. The aqueous layer was extracted twice with ethyl acetate. The combined organic layer was washed with brine and dried with $\mathrm{MgSO}_{4}$. The crude material was purified by silica gel column chromatography using 0-100\% CMA 80/ethyl acetate to yield 70 $\mathrm{mg}(79 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta$ ppm 1.43-1.82 (m, 4 H$) 1.91-2.03(\mathrm{~m}, 1 \mathrm{H}) 2.67(\mathrm{dt}, J=11.68,7.16$ $\mathrm{Hz}, 2 \mathrm{H}) 2.76-2.94(\mathrm{~m}, 1 \mathrm{H}) 3.21(\mathrm{dd}, J=11.82,2.87 \mathrm{~Hz}, 1 \mathrm{H}) 4.28$ (br. s., 1 H$) 6.11(\mathrm{~d}, J=6.12 \mathrm{~Hz}, 1 \mathrm{H}) 6.98-7.57(\mathrm{~m}, 8 \mathrm{H}) 8.36(\mathrm{~s}, 1$ H), $[\mathrm{M}+\mathrm{H}]^{+} 439.6$.
(3R)-3-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-amino\}-N-ethylpiperidine-1-carboxamide (15c). To a solution of $\mathbf{1 5 b}$ ( $20 \mathrm{mg}, 0.046 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.02 \mathrm{~mL}, 0.137 \mathrm{mmol}, 3$ equiv) and ethyl isocyanate $(0.005 \mathrm{~mL}, 0.068 \mathrm{mmol}, 1.5$ equiv). The reaction is stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 15 mg ( $65 \%$ ) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm}$ 1.03-1.34 (m, 3 H ) 1.56-1.96 (m, 4 H) 2.18 (br. s., 1 H) 3.07 (br. s., 2 H) 3.32 (br. s., 1 H) 3.73-4.42 (m, $3 \mathrm{H}) 5.05$ (br. s., 1 H ) 6.02 (d, $J=5.84 \mathrm{~Hz}, 1 \mathrm{H}) 7.09-7.63(\mathrm{~m}, 8 \mathrm{H})$ 8.45 (br. s., 1 H ), $[\mathrm{M}+\mathrm{H}]^{+} 510.3$.

1-[(3R)-3-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-amino\}piperidin-1-yl]ethan-1-one (15d). A solution of $15 b(20 \mathrm{mg}$, $0.046 \mathrm{mmol}, 1$ equiv) in 1 mL of acetic anhydride and 1 mL of pyridine was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 19 mg (87\%) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta$ ppm $1.53-1.92$ (m, 3 H) 2.07 (br. s., 4 H) $2.84-3.37(\mathrm{~m}, 2 \mathrm{H}) 3.86-4.59(\mathrm{~m}, 3 \mathrm{H}) 5.88$ (br. s., 1 H$) 7.04-7.60(\mathrm{~m}, 8 \mathrm{H}) 8.37$ (br. s., 1 H$),[\mathrm{M}+\mathrm{H}]^{+} 481.3$.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)-N-[(3R)-1-methanesulfo-nylpiperidin-3-yl]-9H-purin-6-amine (15e). To a solution of $\mathbf{1 5 b}$ (20 $\mathrm{mg}, 0.046 \mathrm{mmol}$, leq.) in 2 mL of dichloromethane was added methanesulfonyl chloride ( $0.007 \mathrm{~mL}, 0.091 \mathrm{mmol}, 2$ equiv) and triethylamine ( $0.02 \mathrm{~mL}, 0.137 \mathrm{mmol}, 3$ equiv). The mixture was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield $12 \mathrm{mg}(51 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform- $d$ ) $\delta \mathrm{ppm} 1.81-2.18(\mathrm{~m}, 4 \mathrm{H}) 2.77-2.94(\mathrm{~m}, 3$ H) 3.07-3.29 (m, 2 H) 3.42 (br. s., 1 H$) 3.82(\mathrm{~d}, J=9.89 \mathrm{~Hz}, 1 \mathrm{H})$ 4.68 (br. s., 1 H) 6.13 (br. s., 1 H) $7.12-7.65$ (m, 8 H$) 8.48$ (s, 1 H ), $[\mathrm{M}+\mathrm{H}]^{+} 517.8$.
tert-Butyl (3S)-3-\{[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]amino\}piperidine-1-carboxylate (16a). To a solution of 18 ( $100 \mathrm{mg}, 0.267 \mathrm{mmol}, 1$ equiv) in 3 mL of ethanol was added tertbutyl (3S)-3-aminopiperidine-1-carboxylate ( $64 \mathrm{mg}, 0.32 \mathrm{mmol}, 1.2$ equiv) and triethylamine ( $0.06 \mathrm{~mL}, 0.4 \mathrm{mmol}, 1.5$ equiv). The reaction
was heated to $80^{\circ} \mathrm{C}$ for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 137 mg (95\%) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta$ ppm 1.43 (br. s., 9 H ) $1.60-1.76(\mathrm{~m}, 2 \mathrm{H}) 1.77-1.93(\mathrm{~m}, 1 \mathrm{H}) 2.08-2.22(\mathrm{~m}, 1 \mathrm{H}) 3.30$ (br. s., 2 H) 3.60 (br. s., 1 H) 3.98 (br. s., 1 H) 4.42 (br. s., 1 H) 5.96 (d, J $=7.54 \mathrm{~Hz}, 1 \mathrm{H}) 7.14-7.68(\mathrm{~m}, 8 \mathrm{H}) 8.48(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 539.4$.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)-N-[(3S)-piperidin-3-yl]-9H-purin-6-amine (16b). A solution of 16 ( $120 \mathrm{mg}, 0.22 \mathrm{mmol}$ ) was stirred in dichloromethane $(7 \mathrm{~mL})$ and trifluoroacetic acid $(3 \mathrm{~mL})$ for 16 h . The reaction was concentrated in vacuo. The crude product was dissolved in ethyl acetate and washed with saturated $\mathrm{NaHCO}_{3}$. The aqueous layer was extracted twice with ethyl acetate. The combined organic layer was washed with brine and dried with $\mathrm{MgSO}_{4}$. The crude material was purified by silica gel column chromatography using $0-$ $100 \%$ CMA 80/ethyl acetate to yield 64 mg ( $65 \%$ ) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.55-1.86(\mathrm{~m}, 4 \mathrm{H}) 2.01-$ $2.14(\mathrm{~m}, 1 \mathrm{H}) 2.66-2.85(\mathrm{~m}, 2 \mathrm{H}) 2.87-3.04(\mathrm{~m}, 1 \mathrm{H}) 3.32(\mathrm{dd}, J=$ 11.77, $2.35 \mathrm{~Hz}, 1 \mathrm{H}) 4.39$ (br. s., 1 H$) 6.20$ (d, $J=5.84 \mathrm{~Hz}, 1 \mathrm{H})$ 7.12-7.66(m, 8 H) $8.47(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 439.6$.
(3S)-3-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-amino\}-N-ethylpiperidine-1-carboxamide (16c). To a solution of $\mathbf{1 6 b}(16 \mathrm{mg}, 0.036 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.015 \mathrm{~mL}, 0.109 \mathrm{mmol}, 3$ equiv) and ethyl isocyanate ( $0.004 \mathrm{~mL}, 0.055 \mathrm{mmol}, 1.5$ equiv). The reaction is stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield $16 \mathrm{mg}(86 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.06-1.34(\mathrm{~m}, 3 \mathrm{H}) 1.53-1.95(\mathrm{~m}, 4$ H) $2.12-2.29$ (m, 1 H$) 3.07$ (br. s., 2 H$) 3.32$ (br. s., 1 H ) 3.72-4.41 (m, 3 H ) 5.06 (br. s., 1 H$) 6.02(\mathrm{~d}, J=6.12 \mathrm{~Hz}, 1 \mathrm{H}) 7.08-7.66(\mathrm{~m}, 8$ H) 8.45 (br. s., 1 H$),[\mathrm{M}+\mathrm{H}]^{+} 510.3$.

1-[(3S)-3-\{[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-amino\}piperidin-1-yllethan-1-one (16d). A solution of $\mathbf{1 6 b}(15.7 \mathrm{mg}$, $0.036 \mathrm{mmol}, 1$ equiv) in 1 mL of acetic anhydride and 1 mL of pyridine was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 14 mg (81\%) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.48-1.91$ (m, 3 H) $2.07(\mathrm{~s}, 4 \mathrm{H}) 2.90-3.39(\mathrm{~m}, 2 \mathrm{H}) 3.96-4.57(\mathrm{~m}, 3 \mathrm{H}) 5.83(\mathrm{~d}, J=$ $6.50 \mathrm{~Hz}, 1 \mathrm{H}) 7.04-7.56(\mathrm{~m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 481.3$.

8-(2-Chlorophenyl)-9-(4-chlorophenyl)-N-[(3S)-1-methanesulfo-nylpiperidin-3-yl]-9H-purin-6-amine (16e). To a solution of $\mathbf{1 6 b}$ ( $15.3 \mathrm{mg}, 0.035 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added methanesulfonyl chloride ( $0.005 \mathrm{~mL}, 0.07 \mathrm{mmol}, 2$ equiv) and triethylamine ( $0.015 \mathrm{~mL}, 0.11 \mathrm{mmol}, 3$ equiv). The mixture was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield $16 \mathrm{mg}(89 \%)$ of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.63-2.02(\mathrm{~m}, 4 \mathrm{H}$ ) $2.77(\mathrm{~s}, 3 \mathrm{H})$ 2.98-3.19 (m, 2 H) 3.31 (br. s., 1 H ) 3.71 (d, $J=10.55 \mathrm{~Hz}, 1 \mathrm{H}) 4.56$ (br. s., 1 H) $6.01(\mathrm{~d}, J=7.54 \mathrm{~Hz}, 1 \mathrm{H}) 6.97-7.56(\mathrm{~m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1$ $\mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$517.4.
tert-Butyl N-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-$6-y l]$ piperidin-4-ylfcarbamate (17a). To a solution of $18(309 \mathrm{mg}$, $0.824 \mathrm{mmol}, 1$ equiv) in 10 mL of ethanol was added tert-butyl $N$ -(piperidin-4-yl)carbamate (21) ( $329 \mathrm{mg}, 1.65 \mathrm{mmol}, 2$ equiv). The reaction was heated to $80^{\circ} \mathrm{C}$ for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexanes to yield 442 mg (99\%) of desired product. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform- $d$ ) $\delta \mathrm{ppm}$ $1.40-1.53(\mathrm{~m}, 11 \mathrm{H}) 2.12(\mathrm{~d}, J=11.40 \mathrm{~Hz}, 2 \mathrm{H}) 3.32(\mathrm{t}, J=11.63 \mathrm{~Hz}$, 2 H) 3.80 (br. s., 1 H) $4.41-4.68$ (m, 1 H) 5.40 (br. s., 2 H) 7.16$7.24(\mathrm{~m}, 2 \mathrm{H}) 7.26-7.43(\mathrm{~m}, 5 \mathrm{H}) 7.51(\mathrm{~d}, J=6.59 \mathrm{~Hz}, 1 \mathrm{H}) 8.38(\mathrm{~s}, 1$ $\mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$539.2.

1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-amine (17b). A solution of $17 \mathrm{a}(380 \mathrm{mg}, 0.705 \mathrm{mmol})$ was stirred in dichloromethane $(7 \mathrm{~mL})$ and trifluoroacetic acid $(3 \mathrm{~mL})$ for 16 h . The reaction was concentrated in vacuo. The crude product was dissolved in ethyl acetate and washed with saturated $\mathrm{NaHCO}_{3}$.

The aqueous layer was extracted twice with ethyl acetate. The combined organic layer was washed with brine and dried with $\mathrm{MgSO}_{4}$. The crude material was purified by silica gel column chromatography using $0-100 \%$ CMA 80/ethyl acetate to yield 294 mg (95\%) of the desired compound. Compound was determined to be $95 \%$ pure by ${ }^{1} \mathrm{H}$ NMR and was carried forward into further chemistry. ${ }^{1} \mathrm{H}$ NMR (300 MHz , chloroform-d) $\delta \mathrm{ppm} \mathrm{1.31-1.54}(\mathrm{~m}, 2 \mathrm{H}) 1.92-2.08(\mathrm{~m}, 2 \mathrm{H})$ 2.92-3.12 (m, 1 H) $3.27(\mathrm{t}, J=11.87 \mathrm{~Hz}, 2 \mathrm{H}) 5.42$ (br. s., 2 H$)$ $7.07-7.61(\mathrm{~m}, 8 \mathrm{H}) 8.38(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 439.4$.

General Procedure for Making Ureas from 1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]piperidin-4-amine (17b). To a solution of $\mathbf{1 7 b}(20 \mathrm{mg}, 0.046 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.02 \mathrm{~mL}, 0.136 \mathrm{mmol}, 3$ equiv) and the appropriate isocyanate ( 1.5 equiv). The reaction was stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexane to yield pure compound.

1-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-y/\}-3-ethylurea (17c). Reaction proceeded in $56 \%$ yield. ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, chloroform- $d) \delta \mathrm{ppm} 1.13(\mathrm{t}, J=7.21 \mathrm{~Hz}, 3 \mathrm{H})$ $1.33-1.57(\mathrm{~m}, 2 \mathrm{H}) 2.04-2.20(\mathrm{~m}, 2 \mathrm{H}) 3.14-3.25(\mathrm{~m}, 2 \mathrm{H}) 3.32(\mathrm{t}, J$ $=12.06 \mathrm{~Hz}, 2 \mathrm{H}) 3.86-4.05(\mathrm{~m}, 1 \mathrm{H}) 4.23-4.45(\mathrm{~m}, 2 \mathrm{H}) 5.39$ (br. s., $2 \mathrm{H}) 7.03-7.59(\mathrm{~m}, 8 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 510.2$.

1-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}-3-(propan-2-yl)urea (17d). Reaction proceeded in $84 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.14$ (d, $J=$ $6.50 \mathrm{~Hz}, 6 \mathrm{H}) 1.35-1.55(\mathrm{~m}, 2 \mathrm{H}) 2.12(\mathrm{~d}, J=10.46 \mathrm{~Hz}, 2 \mathrm{H}) 3.32(\mathrm{t}$, $J=12.10 \mathrm{~Hz}, 2 \mathrm{H}) 3.84(\mathrm{dd}, J=13.66,6.69 \mathrm{~Hz}, 1 \mathrm{H}) 3.89-4.04(\mathrm{~m}, 1$ H) $4.12-4.36(\mathrm{~m}, 2 \mathrm{H}) 5.39$ (br. s., 2 H$) 7.19(\mathrm{~d}, J=8.67 \mathrm{~Hz}, 2 \mathrm{H})$ $7.28-7.43(\mathrm{~m}, 5 \mathrm{H}) 7.51(\mathrm{~d}, J=6.69 \mathrm{~Hz}, 1 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+$ $\mathrm{H}]^{+} 524.6$.

1-\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}-3-propylurea (17e). Reaction proceeded in $71 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.83-0.98$ (m, 3 H ) 1.37$1.59(\mathrm{~m}, 4 \mathrm{H}) 2.13(\mathrm{~d}, J=10.27 \mathrm{~Hz}, 2 \mathrm{H}) 3.12(\mathrm{q}, J=6.66 \mathrm{~Hz}, 2 \mathrm{H})$ $3.32(\mathrm{t}, J=12.01 \mathrm{~Hz}, 2 \mathrm{H}) 3.83-4.07(\mathrm{~m}, 1 \mathrm{H}) 4.20-4.47(\mathrm{~m}, 2 \mathrm{H})$ 5.40 (br. s., 2 H ) $7.13-7.24(\mathrm{~m}, 2 \mathrm{H}) 7.28-7.44(\mathrm{~m}, 5 \mathrm{H}) 7.51(\mathrm{~d}, \mathrm{~J}=$ 6.69 Hz, 1 H) $8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$524.1.

3-Butyl-1-\{1-[8-(2-chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]piperidin-4-yl\}urea (17f). Reaction proceeded in $69 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 0.84-0.99(\mathrm{~m}, 3 \mathrm{H}) 1.28-$ $1.55(\mathrm{~m}, 6 \mathrm{H}) 2.04-2.20(\mathrm{~m}, 2 \mathrm{H}) 3.15(\mathrm{q}, J=6.75 \mathrm{~Hz}, 2 \mathrm{H}) 3.32(\mathrm{t}, J$ $=12.15 \mathrm{~Hz}, 2 \mathrm{H}) 3.83-4.08(\mathrm{~m}, 1 \mathrm{H}) 4.18-4.39(\mathrm{~m}, 2 \mathrm{H}) 5.40$ (br. s., $2 \mathrm{H}) 7.08-7.43(\mathrm{~m}, 7 \mathrm{H}) 7.51(\mathrm{~d}, J=6.69 \mathrm{~Hz}, 1 \mathrm{H}) 8.37(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}$ $+\mathrm{H}]^{+} 538.4$.

General Procedure for Making Amides from 1-[8-(2-Chlorophen-yl)-9-(4-chlorophenyl)-9H-purin-6-yllpiperidin-4-amine (17b). To a solution of $\mathbf{1 7 b}(21 \mathrm{mg}, 0.048 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine $(0.02 \mathrm{~mL}, 0.143 \mathrm{mmol}, 3$ equiv) and the appropriate anhydride ( 2 equiv). The reaction is stirred for 16 h . The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using 0-100\% ethyl acetate/hexane to yield pure compound.
$N$-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}acetamide (17g). Reaction proceeded in $65 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.42-1.56$ (m, 2 H) 1.99 (s, 3 H) $2.12(\mathrm{~d}, J=10.08 \mathrm{~Hz}, 2 \mathrm{H}) 3.30(\mathrm{t}, J=12.24 \mathrm{~Hz}, 2 \mathrm{H}) 4.09-4.25$ $(\mathrm{m}, 1 \mathrm{H}) 5.41(\mathrm{~d}, J=8.01 \mathrm{~Hz}, 3 \mathrm{H}) 7.13-7.43(\mathrm{~m}, 7 \mathrm{H}) 7.51(\mathrm{~d}, J=$ $6.78 \mathrm{~Hz}, 1 \mathrm{H}) 8.38(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 481.4$.

N-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}-2,2,2-trifluoroacetamide (17i). Reaction proceeded in $39 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.62$ (qd, $J=$ $11.99,3.86 \mathrm{~Hz}, 2 \mathrm{H}) 2.18(\mathrm{~d}, J=10.83 \mathrm{~Hz}, 2 \mathrm{H}) 3.31(\mathrm{t}, J=12.43 \mathrm{~Hz}$, $2 \mathrm{H}) 4.11-4.29(\mathrm{~m}, 1 \mathrm{H}) 5.53$ (br. s., 2 H$) 6.19(\mathrm{~d}, J=7.06 \mathrm{~Hz}, 1 \mathrm{H})$ $7.01-7.57(\mathrm{~m}, 8 \mathrm{H}) 8.40(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 535.4$.
$N$-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}benzamide (17j). Reaction proceeded in $19 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.82-1.97$ (m, 2 H ) 2.77 (d, J $=12.34 \mathrm{~Hz}, 2 \mathrm{H}) 3.08$ (br. s., 2 H ) 3.46-3.62(m, 1 H) 3.66-3.80 (m, $1 \mathrm{H}) 4.01-4.13(\mathrm{~m}, 1 \mathrm{H}) 5.26-5.46(\mathrm{~m}, 1 \mathrm{H}) 7.12-7.59(\mathrm{~m}, 13 \mathrm{H})$ $8.38(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+} 543.6$.

General Procedure for Making Sulfonamides from 1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]piperidin-4-amine (17b). To a solution of $17 \mathrm{~b}(21 \mathrm{mg}, 0.048 \mathrm{mmol}, 1$ equiv) in 2 mL of THF was added triethylamine ( $0.02 \mathrm{~mL}, 0.143 \mathrm{mmol}, 3$ equiv) and the appropriate sulfonyl chloride ( 2 equiv). The reaction is stirred for 16 $h$. The reaction was concentrated in vacuo. The crude material was purified by silica gel column chromatography using $0-100 \%$ ethyl acetate/hexane to yield pure compound.

N-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}methanesulfonamide (17h). Reaction proceeded in $32 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.55-1.67$ (m, $2 \mathrm{H}) 2.18(\mathrm{~d}, \mathrm{~J}=12.53 \mathrm{~Hz}, 2 \mathrm{H}) 3.03(\mathrm{~s}, 3 \mathrm{H}) 3.36(\mathrm{t}, J=12.29 \mathrm{~Hz}, 2$ H) $3.59-3.80(\mathrm{~m}, 1 \mathrm{H}) 4.31(\mathrm{~d}, J=7.44 \mathrm{~Hz}, 1 \mathrm{H}) 5.43(\mathrm{~d}, J=9.89$ $\mathrm{Hz}, 2 \mathrm{H}) 7.13-7.44(\mathrm{~m}, 7 \mathrm{H}) 7.51(\mathrm{~d}, J=6.78 \mathrm{~Hz}, 1 \mathrm{H}) 8.39(\mathrm{~s}, 1 \mathrm{H})$, $[\mathrm{M}+\mathrm{H}]^{+}$517.6.

N-\{1-[8-(2-Chlorophenyl)-9-(4-chlorophenyl)-9H-purin-6-yl]-piperidin-4-yl\}benzenesulfonamide (17k). Reaction proceeded in $53 \%$ yield. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , chloroform-d) $\delta \mathrm{ppm} 1.43-1.59$ (m, $2 \mathrm{H}) 1.94(\mathrm{~d}, J=10.46 \mathrm{~Hz}, 2 \mathrm{H}) 3.30(\mathrm{t}, J=11.96 \mathrm{~Hz}, 2 \mathrm{H}) 3.41-3.66$ $(\mathrm{m}, 1 \mathrm{H}) 4.62(\mathrm{~d}, J=7.54 \mathrm{~Hz}, 1 \mathrm{H}) 5.26(\mathrm{~d}, J=10.08 \mathrm{~Hz}, 2 \mathrm{H}) 7.07-$ $7.69(\mathrm{~m}, 10 \mathrm{H}) 7.92(\mathrm{~d}, J=7.35 \mathrm{~Hz}, 2 \mathrm{H}) 8.35(\mathrm{~s}, 1 \mathrm{H}),[\mathrm{M}+\mathrm{H}]^{+}$ 579.4.

Calcium Mobilization and Radioligand Displacement Assays. Each compound was pharmacologically characterized using a functional fluorescent CB1 activated $\mathrm{G} \alpha \mathrm{q}_{16}$-coupled intracellular calcium mobilization assay in CHO-K1 cells, as has been described in our previous publications and apparent affinity $\left(K_{\mathrm{e}}\right)$ values were determined. ${ }^{13}$ Briefly, CHO-K1 cells were engineered to coexpress human CB1 and $\mathrm{G} \alpha \mathrm{q}_{16}$. Activation of CB1 by an agonist then leads to generation of inositol phospahatase 3 (IP3) and activation of IP3 receptors, which leads to mobilization of intracellular calcium. Calcium flux was monitored in a 96 -well format using the fluorescent dye Calcein-4 a.m. in an automated platereader (Flexstation, Molecular Devices). The antagonism of a test compound was measured by its ability to shift the concentration response curve of the synthetic CB1 agonist CP55940 rightwards using the equation:
$K_{\mathrm{e}}=$ [Ligand] $/[\mathrm{DR}-1]$ where DR is the $\mathrm{EC}_{50}$ ratio of CP55940 in the presence or absence of a test agent. Standard errors were between 5 and $25 \%$ of mean in most cases and have been left out for clarity.

Further characterization of select compounds was performed using radioligand displacement of $\left[{ }^{3} \mathrm{H}\right]$ SR141716 and equilibrium dissociation constant $\left(K_{\mathrm{i}}\right)$ values were determined as described previously. ${ }^{13}$ Selectivity of these compounds at CB1 versus CB2 was also determined by obtaining $K_{\mathrm{i}}$ values at either receptor using displacement of $\left[{ }^{3} \mathrm{H}\right]$ CP55940 in membranes of CHO-K1 cells overexpressing either receptor. Data reported are average values from 3 to 6 measurements. The standard errors for most measurements were between 5 and $25 \%$ of mean and have been left out from the tables and figures for clarity.

MDCK-mdr1 Permeability Assays. MDCK-mdr1 cells permeability assays were preformed as previously described. ${ }^{11 \mathrm{~b}, 13}$ MDCKmdrl cells obtained from Netherlands Cancer Institute were grown on Transwell type filters (Corning) for 4 days to confluence in DMEM/ F12 media containing $10 \%$ fetal bovine serum and antibiotics as has been described previously. ${ }^{13}$ Compounds were added to the apical side at a concentration of $10 \mu \mathrm{M}$ in a transport buffer comprising of $1 \times$ Hank's balanced salt solution, 25 mM d-glucose and buffered with HEPES to pH 7.4. Samples were incubated for 1 h at $37^{\circ} \mathrm{C}$ and carefully collected from both the apical and basal side of the filters. Compounds selected for MDCK-mdrl cell assays were infused on an Applied Biosystems API-4000 mass spectrometer to optimize for analysis using multiple reaction monitoring (MRM). Flow injection analysis was also conducted to optimize for mass spectrometer parameters. Samples from the apical and basolateral side of the MDCK cell assay were dried under nitrogen on a Turbovap LV. The chromatography was conducted with an Agilent 1100 binary pump with a flow rate of $0.5 \mathrm{~mL} / \mathrm{min}$. Mobile phase solvents were A, $0.1 \%$ formic acid in water, and B, $0.1 \%$ formic acid in methanol. The initial solvent conditions were $10 \%$ B for 1 min , then a gradient was used by
increasing to $95 \%$ B over 5 min , then returning to initial conditions. Data reported are average values from 2 to 3 measurements.

In Vitro Stability Testing. Stability of compounds to plasma and S9 fraction was preformed as previously described. ${ }^{13}$ In vitro testing for metabolic stability was conducted in pooled samples of mixed gender human plasma from BioChemed Services, Winchester, VA and human mixed gender pooled hepatic S9 fraction supplied by Xenotech, LLC, Lenexa, KS. Identity of the donors was unknown.

For the hepatic S 9 metabolism studies, all samples were tested at 10 $\mu \mathrm{M}$ final concentration in a 1 mL volume containing $1 \mathrm{mg} / \mathrm{mL}$ S9. Samples were incubated in a buffer containing 50 mM potassium phosphate, pH 7.4 with $3 \mathrm{mM} \mathrm{MgCl}_{2}$ and a NADPH regeration system comprising of NADP ( 1 mM ), glucose-6-phosphate ( 5 mM ) and glucose-6-phosphate dehydrogenase ( 1 unit $/ \mathrm{mL}$ ). Triplicate samples were incubated for $0,15,30,60$, and 120 min . Reactions were terminated by addition of 3 vol of acetonitrile and processed as described for the MDCK-mdr1 assays, but standard curves were prepared in blank matrix for each compound for quantitative assessment.

The plasma stability studies were conducted at $37{ }^{\circ} \mathrm{C}$ in a volume of 1 mL plasma per sample. All compounds were tested at $10 \mu \mathrm{M}$ final concentration at 0,30 , and 60 min after a 5 min preincubation. Reactions were terminated by addition of acetonitrile and analyzed as described above.

Pharmacokinetic Evaluation in Vivo. Male Sprague-Dawley rats aged 8 weeks at time of dosing were acquired from Charles River Laboratories and were dosed orally. Three to four animals were used per time point. Dose was formulated in food-grade corn oil at 10 mg / kg. Plasma, liver, and brain were taken from all rats at $0.5,1,2,4$, and 8 h postdose. Samples were prepared and analyzed as follows: Plasma ( $50 \mu \mathrm{~L}$ ) was mixed with $10 \mu \mathrm{~L}$ of $10 \mu \mathrm{~L}$ of acetonitrile, and $300 \mu \mathrm{~L}$ of acetonitrile, vortexed, and centrifuged at 9000 g for 5 min . Supernatants were transferred to autosampler vials with low-volume inserts and injected without dilution. Brains were homogenized with 50:50 ethanol/water (3:1, v/v) using a Potter Elvehjem type homogenizer. Homogenate ( $50 \mu \mathrm{~L}$ ) was mixed with $10 \mu \mathrm{~L}$ of acetonitrile, and 300 $\mu \mathrm{L}$ of acetonitrile, vortexed and centrifuged at 9000 g for 5 min . Supernatant was transferred to autosampler vials with low-volume inserts and injected without dilution. Standards were prepared as above for each compound in blank plasma, blank liver homogenate, and blank brain homogenate. Standards used were within $15 \%$ of nominal; except for $20 \%$ at LOQ. Compounds for LC-MS/MS analyses were supplied at $1 \mathrm{mg} / \mathrm{mL}$ in methanol. The stock solutions were further diluted to $\sim 100 \mathrm{ng} / \mathrm{mL}$. The $100 \mathrm{ng} / \mathrm{mL}$ solutions were used to optimize the mass spectrometer for MRM transitions and mass spectrometer parameters. Infusion and flow injection optimization were also performed. LC-MS/MS was conducted using an Applied Biosystems API 4000 coupled with an Agilent 1100 HPLC system. Chromatography was performed with a Phenomenex Luna C18 (50 $\times$ $2 \mathrm{~mm}, 5 \mu \mathrm{~m}$ ) column. Mobile phases were $0.1 \%$ formic acid and 10 mM ammonium formate in water (A), and $0.1 \%$ formice acid and 10 mM ammonium formate in methanol. Initial conditions were $10 \% \mathrm{~B}$ and held for one minute, followed by a linear gradient to $90 \%$ B over 5 $\mathrm{min} .90 \% \mathrm{~B}$ was held for 2 min before returning to initial conditions. Compound 12 was analyzed with multiple reaction monitoring in the positive mode with a transition of $595.28 \rightarrow$ 370.0. The following parameters were used, $\mathrm{DP}=116, \mathrm{CE}=45, \mathrm{CXP}=36, \mathrm{CAD}=4, \mathrm{CUR}$ $=10, \mathrm{GS} 1=40, \mathrm{GS} 2=60, \mathrm{IS}=4000$, and $\mathrm{TEM}=650$.

## ASSOCIATED CONTENT

## (s) Supporting Information

HPLC and melting point data of target compounds. This material is available free of charge via the Internet at http:// pubs.acs.org.

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## Notes

The authors declare no competing financial interest.

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## ABBREVIATIONS USED

CB1, cannabinoid receptor 1; CB2, cannabinoid receptor 2 ; CMA 80, $80 \%$ chloroform, $18 \%$ methanol, and $2 \%$ ammonium hydroxide; CNS, central nervous system; BBB, blood-brain barrier; TPSA, topological polar surface area; ECS, endocannabinoid system; CBR, cannabinoid receptors; $K_{\text {e }}$ apparent affinity constant, MDCK-mdr1, Madin-Darby canine kidney cells transfected with the human MDR1 gene; A, apical; B, basal; BOP, benzotriazole-1-yl-oxytris(dimethylamino)phosphonium hexafluorophosphate; CHO-K1, Chinese hamster ovary cells; $\mathrm{IP}_{3}$, inositol phospahatase 3 ; MRM, multiple reaction monitoring; LOQ below limit of quantitation; NA, not applicable

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[^1]:    ${ }^{a}$ Compound's permeability was measured from apical to basal (A to B) sides of the membrane.

